UV Raman spectroscopy of group IV nanocrystals embedded in a SiO₂ matrix

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Abstract Nanostructures of both Ge nanocrystals formed by thermal oxidation of SiGe layers, and SiGe nanocrystals formed by crystallization of amorphous SiGe nanoparticles deposited by LPCVD have been analyzed by Raman spectroscopy. The nanostructures are formed on a silicon substrate. Raman spectra have been acquired with visible (514.5 nm) and UV (325 nm) excitation lines. When the amount of material is very small, as it has happens in these nanostructures, the visible line is not able to excite the characteristic peaks of the Ge or SiGe in the Raman spectrum; instead the Si second order spectrum of the substrate appears and it can be misinterpreted by attributing it to the Ge-Ge band associated with the nanocrystals. In this work, the use of UV excitation has been demonstrated to enhance the sensitivity respect to the conventional visible excitation, allowing the characteristic peaks of the Ge or SiGe nanocrystals to appear in the spectrum. We attributed this effect to the resonance effects.

1 Introduction

Group IV semiconductor nanocrystals embedded in a dielectric matrix have received a great interest for their potential application in Si based optoelectronics and non-volatile memories. The optical properties of Si and Ge are improved by reducing the particle size to the nanoscale. The band gap of these nanocrystals is determined by the nanocrystal size. An additional tunability tool is the use of SiGe alloys with varying composition. The fabrication of these nanostructures has been undertaken by means of different procedures. Among other, oxidation of polycrystalline SiGe layers and crystallization of amorphous SiGe nanoparticles embedded in SiO₂ deposited by Low Pressure Chemical Vapour Deposition (LPCVD) are methods technologically simple and fully compatible with the CMOS technology. However, there are several technical barriers that mainly concern the control of the composition of the nanoparticles during the treatments leading to the formation of the nanocrystals. On the other hand, the crystallinity of the nanoparticles, their size and the residual stresses are factors that affect in a significant way the properties of the nanostructures. Therefore, a full characterization is necessary in order to control the main characteristics of the nanocrystals embedded in the SiO₂ matrix for specific applications. Raman spectroscopy appears as a suitable tool to characterize these structures, because it is non-destructive and is sensitive to most of the structural features mentioned above.

These nanostructures with nanocrystals embedded in the dielectric matrix are usually grown on a Si substrate; in this cases the Raman spectrum is a mixture of the ones of the nanostructure and the Si substrate; therefore, the contribution of the nanocrystals to the Raman spectrum has to be separated from the substrate contribution. In general, there is a very small amount of matter in these nanostructures; even some applications demand a few nanocrystals, which makes very difficult the detection of the Raman signal arising from them. In this case, the Raman signal of the nanocrystals can be washed out by the second order spectrum arising from the Si substrate.
The Raman intensity depends on the Raman scattering cross section and the scattering volume, which in the case of the nanocrystals is mainly determined by the amount of material, rather than by the absorption coefficient as for bulk samples; in fact, the signal arising from the Si substrate emerges in the Raman spectra obtained on these nanostructures, which means that the stacked thickness of nanocrystals is smaller than the Raman probe depth 1/2x (with x the absorption coefficient at the excitation wavelength).

The use of UV excitation is demonstrated in this work to enhance the sensitivity respect to the conventional visible excitation, allowing the detection of the Raman spectrum of very small amounts of nanocrystals.

2 Experimental procedures

2.1 Sample preparation

We present herein the study of nanostructures formed by two different procedures: oxidation of SiGe layers, and thermal crystallization of amorphous SiGe nanoparticles.

Ge nanocrystals embedded in SiO₂ were prepared as follows. Amorphous SiGe layers were deposited by LPCVD in a hot wall reactor at 425 °C. The pressure during growth was 300 mTorr and the flow rates of the precursor gases (Si₂H₆ and GeH₄, without carrier gas) were adjusted to obtain layers with Ge fraction x = 0.3. The substrates were silicon wafers previously oxidized to a SiO₂ thickness of 750 nm. The films were annealed at 725 °C for 72 h, which is a time sufficiently long to complete the crystallization. The annealing was carried out in vacuum (pressure below 60 mTorr) to avoid the uncontrolled oxidation of the samples; for more details see By means of this procedure, a set of samples of continuous 50 nm thick amorphous SiGe layer was thermally crystallized, and subsequently oxidized in a dry oxygen atmosphere at 850 °C for times increasing up to 7 h. Phase separation constituted a barrier to control the particle composition, which did not keep the starting composition then becoming Ge-rich and evolving to pure Ge nanocrystals.

SiGe nanocrystals embedded in SiO₂ were obtained as follows. The structures, deposited by LPCVD in a single run at 390 °C, consisted of a bottom SiO₂ layer deposited onto a Si (001) substrate, a SiGe layer with a Ge fraction of around 0.4 and a top SiO₂ capping layer. SiGe is deposited using GeH₄ and Si₂H₆ as precursor gases. SiO₂ is deposited from Si₂H₆ and O₂ using a flow ratio of 0.2. Samples with SiGe layers of different thickness were deposited by adjusting the deposition time. For thickness of 5 nm or higher, the SiGe layers are continuous, but layers below 5 nm are discontinuous giving rise to nanoparticles. Three samples have been deposited, one sample with a continuous SiGe layer 15 nm thick, another sample with a quasi continuous layer with nanoparticles of 5 nm diameter, and another one with around 3.5 nm diameter nanoparticles. The samples were annealed by rapid thermal annealing (RTA) in N₂ atmosphere, at 800 °C for 1 min. These procedures are described elsewhere For more details about the crystallization of small nanoparticles of either Si or Ge

2.2 Characterization by Raman spectroscopy

The Raman spectra were acquired with both visible (514.5 nm line from an Ar + laser) and UV (325 nm line from a HeCd laser) excitation using a HR UV Labram Jobin–Yvon Raman spectrometer. The results obtained with visible and UV excitations in thick layers were substantially identical. However, for nanostructures with very small particle size the UV excitation presented significant improvement respect to the visible excitation, as discussed later on. The Raman spectra of crystalline group IV semiconductors, Si and Ge, mainly consist of a first order phonon band, 520 cm⁻¹ for Si, and 302 cm⁻¹ for Ge. In the case of SiGe alloys first order phonon modes, arising from the vibrations associated with the different bonds in the alloy, Si–Si, Ge–Ge and Si–Ge, are detected. The spectral features of these modes provide information about the chemical composition of the alloy These phonon bands are very sensitive to the crystal environment, in particular the loss of translation symmetry induces broadening and red frequency shifts, which are used to determine the crystallite size Also, strain induces a frequency shift, which is positive under compressive strain, and negative under tensile strain Besides the first order phonon bands, a series of second order (two phonon processes) bands, much weaker than the first order bands, is observed.

3 Results and discussion

Figure 1 presents the second order Raman spectrum of Si with the 2TA (transverse acoustic) band peaking at 300 cm⁻¹, and the first order Raman spectrum of Ge nanocrystals of the samples obtained by thermal oxidation of SiGe layers both acquired under visible excitation. The two spectra show a strong similitude, which can induce to misleading interpretations when the amount of Ge nanocrystals is quite small. This problem was addressed by A.V. Kobolov , who showed the inconsistency of many Raman spectra reported as associated with Ge nanocrystals, as in reality they arise from the Si substrate. Furthermore,
Kolobov asked himself about the real possibility of measuring the Raman spectrum of Ge nanostructures on Si substrates. One possibility to do it is to enhance the Raman signal by resonance.

As seen in Fig. 2 the imaginary part of the dielectric function presents three resonances for Ge at 2.3 eV, 3.2 eV and 4.2 eV, and two resonances at 3.4 eV and 4.2 eV for Si

The Raman scattering cross section is enhanced by resonance at energies close to the optical transitions for polar phonon modes. Furthermore, the absorption coefficient of Ge is strongly increased for UV excitation which also contributes to enhance the Raman signal. The above resonances are valid for bulk material. In the case of nanocrystals, quantum confinement shifts the energy of the optical transitions; therefore, one can expect resonances at different energy values. The resonance conditions should depend on the crystal size, which makes UV excitation suitable for increasing the intensity of the Raman signal of small nanocrystals.

Resonance effects under UV excitation were observed for Ge nanocrystals embedded in SiO$_2$ formed by oxidation of SiGe polycrystalline layers. In this samples, once all the Si atoms of the SiGe layer have been oxidized a continuous thin Ge layer is formed after an oxidation time of 225 min; further oxidation thins down the layer and forms Ge islands, which the size is progressively reduced by oxidation forming Ge nanocrystals of progressively smaller diameter.

Figure 3 shows the Raman intensity of the Ge nanocrystals as a function of the oxidation time for both excitations (514.5 nm and 325 nm). The intensities are normalized to the intensity measured for the continuous Ge layer. The evolution of the Raman intensity shows the existence of a resonance for UV excitation; the Raman intensity increases for decreasing nanocrystal size. Note that indeed the Raman intensity increase should be even higher, because the volume occupied by the Ge nanocrystals is reduced for increasing oxidation time. Contrarily to the evolution of the Raman intensity for UV excitation, the Raman intensity for visible excitation does not show any significant evolution with the nanocrystal size, which is in agreement with the small confinement energy reported for the E1 excitonic transition for Ge nanocrystals with size around 5 nm the E1 gap scales from 2.2 to 3.3 eV for nanocrystals with 2 nm diameter. The average of the nanocrystals size in our samples is shown in Fig. 3. The decrease of the intensity for both excitations for long oxidation times, 315 min, obeys the drastic reduction of the Ge volume at such an oxidation stage. The resonance associated with the UV excitation does not seem to be
related to E 1 resonance, but it must be associated with higher energy transitions, either the zone center E' 0 direct transition (3.2 eV in bulk Ge) or the zone border E 2 transition (4.3 eV).

The average nanocrystal size was estimated by fitting the Raman spectra using a phonon correlation length model (Fig. 4). The optimal fit to the experimental data (triangles) is achieved when the compressive strain is considered, solid line, while it is shifted to the low frequency respect to the experimental plot when only phonon confinement is taken into account, dashed line. A description of the fitting procedure including strain can be found in the text.

This compressive strain supplies an additional contribution to the blue shift of the band gaps.

Figure 5 shows the Raman spectra of SiGe nanocrystals, around 3.5 nm in diameter, embedded in SiO 2 on a Si substrate, obtained with either visible (514.5 nm) or UV (325 nm) excitation; the second order Raman spectrum (visible excitation) of a bare Si substrate is also shown. If one compares the visible spectrum of the SiGe nanostucture and the Si substrate spectrum, one observes that they have an one to one correspondence, evidencing that visible excitation cannot detect the small amount of SiGe nanocrystals. The spectrum of Fig. 5 obtained under UV excitation.

Fig. 4 Experimental Raman spectrum of Ge nanocrystals (triangles). Theoretical curve calculated using the phonon correlation length model (dashed lines). This curve is shifted to high frequencies (continuous line) when the compressive strain is taken into account, fitting the experimental data.

Fig. 5 Raman spectra of SiGe nanocrystals with UV (1) and visible (2) excitation lines respectively. The second order spectrum of the bare Si substrate is also drawn (3).

Fig. 6 Raman spectra of samples with SiGe layers of different thickness (see text). (a) Visible excitation; (b) UV excitation.
tation, clearly shows the Ge–Ge and Si–Ge bands corresponding to the presence of the SiGe nanocrystals.

Figure 6 shows the Raman spectra of the three SiGe samples with visible and UV excitation. The spectra obtained with visible excitation, Fig. 6a, reveal the presence of SiGe only for the 15 nm thick layer, while the spectra of the samples with nanoparticles, 5 and 3.5 nm in diameter, cannot reveal their presence. The only features observed correspond to the first and second order spectrum of the Si substrate. Nevertheless, the spectra obtained with UV excitation, Fig. 6b, show for the three samples the Ge–Ge and Si–Ge bands corresponding to the SiGe nanoparticles, suggesting that the 325 nm (3.81 eV) excitation is close to a resonance of the SiGe nanocrystal system. As it can be seen in Fig. 2, the SiGe resonances must appear in a rough estimation in between 3.5 and 4.5 eV.

In summary, the use of the UV (325 nm) instead of visible (514.5 nm) excitation allows the study by Raman spectroscopy of nanostructures with a very small amount of nanocrystalline material due to the increase in sensitivity caused by the resonance effects. This effect enhances the Ge–Ge and Si–Ge bands of the spectrum of the nanoparticles and therefore avoids possible misleading interpretations caused by the influence of the second order spectrum of the Si substrate.

References