



1 **Keywords:**

2 Mechanical recycling; Poly(lactic acid); Nanocomposites; Structure; Properties

3

4 **1. Introduction**

5 Poly(lactic acid) (PLA) represents an interesting alternative to the fossil-fuel  
6 based polymers in packaging applications, especially for food products. This is due to  
7 PLA being considered as biodegradable, safe in food contact and comparable to some  
8 commodity plastics, such as polyethylene terephthalate (PET), in optical and mechanical  
9 properties [1-3] . However, other PLA properties, such as impact strength and gas  
10 permeability, show only moderate values and therefore the addition of different  
11 nanometric reinforcing agents, such as organically modified clays, has been studied over  
12 the past years. The nanocomposites obtained show, in general, improved impact  
13 strength, fire resistance and gas barrier properties [4-6]. These characteristics have  
14 placed PLA as one of the most important bioplastics in the market, with a global  
15 production capacity of 205000 t in 2014, which is expected to exceed 500000 t in 2020  
16 [7].

17 The expected growth in production of PLA could generate some environmental  
18 and social issues, especially if the origin of the raw materials necessary for its production  
19 is taken into account. PLA is produced by the ring opening polymerization of lactide, the  
20 cyclic dimer of lactic acid, which is obtained by the fermentation of the glucose present  
21 in some food products such as corn, potato or sugar beet [8]. This situation could cause  
22 an increasing demand of large amounts of land to produce such food products, also  
23 increasing the environmental impact of the manufacture of PLA [9]. Furthermore, some  
24 farmers could be tempted to exchange the cheap food production for the crops used in  
25 the production of PLA, compromising the health and sustenance of developing countries  
26 [10]. Lastly, there are some concerns regarding the end-of-life scenarios for this polymer,  
27 since they play a very important role on the environmental impact of PLA. Some studies,  
28 such as those conducted by Piemonte [11], Cosate de Andrade et al. [12] and Rossi et  
29 al. [13] point out that composting is not necessarily the best alternative in the case of  
30 PLA, suggesting that mechanical recycling might be a more interesting alternative. This  
31 is especially important in the commercial grades used in packaging applications, since  
32 they degrade at a slower rate than the accumulation of wastes [14-16].

33 Considering the social and environmental issues derived from the massive  
34 production of PLA and PLA-based materials, it can be concluded that mechanical  
35 recycling is an interesting end-of-life scenario for PLA wastes coming from packaging  
36 applications, since it allows the reduction in raw materials, energy and emissions,, thus  
37 decreasing the environmental impact of the use of PLA. However, in order to evaluate  
38 the feasibility of the mechanical recycling, it is necessary to compare the properties of  
39 virgin and recycled materials, since if there is an important decrease in the properties of  
40 the recycled materials, the mechanical recycling would be unfeasible.

41 Despite the clear interest of the mechanical recycling, the properties of recycled  
42 PLA and PLA-based materials have been scarcely studied up to the present. Different  
43 authors have measured the thermal and mechanical properties of neat PLA after several  
44 reprocessing steps. For instance, Scaffaro et al. [17], Badía et al. [18] and Żenkiewicz et  
45 al [19] reported that PLA presented reduced mechanical properties after 1, 5 and 10

1 reprocessing cycles. However, Nascimento et al. [20] pointed out that a single  
2 reprocessing step does not significantly affect the structure, thermal and mechanical  
3 properties of PLA.

4 In general, these studies do not consider the degradation (thermal,  
5 photochemical and hydrolytic) of the PLA during its useful life nor do stages of  
6 accelerated aging, although the degradation of the polymer prior to reprocessing can  
7 play a very important role in the decrease of the properties of the recycled plastic. In a  
8 previous work we have studied the hydrolytic degradation of PLA subjected to different  
9 recycling processes that include, in addition to the reprocessing step, accelerated aging  
10 and washing steps to simulate the degradation of the material during the service life and  
11 the cleaning of a plastic waste coming from food packaging, respectively [21]. The results  
12 revealed that the recycled plastics show good resistance against the hydrolytic  
13 degradation, although the behavior of the recycled materials depends on the conditions  
14 of the recycling process. The impact of the different mechanical recycling processes on  
15 the mechanical, optical, thermal and barrier properties of the recycled plastic was also  
16 limited [1].

17 As it has been mentioned, the mechanical recycling of unfilled PLA has already  
18 been studied and reported in the literature. However, the behavior of PLA based  
19 nanocomposites does not have to be the same, since the presence of clays could  
20 promote the degradation of the polymer during its service life and also during the  
21 recycling processes, thus affecting the final properties of the recycled materials. Despite  
22 the growing use of clays in PLA based materials, in order to improve the properties of  
23 the polymer, there is little data available regarding the behavior of mechanically recycled  
24 PLA nanocomposites. In this regard, Kozłowski and Macyszyn studied the reprocessing  
25 of a PLA-montmorillonite nanocomposite and found that PLA suffered increasing  
26 degradation during the successive reprocessing cycles, which led to higher values  
27 of the melt flow rate. However, oxygen permeability was reduced as a result of the  
28 reprocessing [22]. In the same vein, Scaffaro et al. studied the effect of five reprocessing  
29 cycles in nanocomposites of PLA and hydrotalcites. They reported that reprocessing  
30 caused an improvement of the dispersion of the clay particles, which led to an increase  
31 of the mechanical properties in the first three reprocessing steps. However, after four  
32 and five reprocessing cycles, the degradation of PLA was severe, thus decreasing the  
33 performance of the material [23]. In both cited works, the degradation of the material  
34 during its service life and possible washing steps were not considered.

35 Consequently, the main aim of this work was to study the effect of different  
36 mechanical recycling processes, including accelerated aging stages, on the properties  
37 of PLA-clay nanocomposites. A commercial grade of PLA, especially designed for  
38 packaging applications, and 2% wt. of an organically modified montmorillonite were melt  
39 compounded and compression molded into films. A portion of these films were subjected  
40 to two different mechanical recycling processes, one comprising an accelerated thermal  
41 and photochemical aging before the second extrusion and compression molding step,  
42 and other including a demanding washing step between the accelerated aging and the  
43 reprocessing step. The effect of both processes on the structure and properties of the  
44 material was followed by water absorption measurements, dilute solution viscosimetry  
45 (IV), infrared spectroscopy (FTIR-ATR), X-ray diffraction (XRD), transmission electron  
46 microscopy (TEM), differential scanning calorimetry (DSC), thermogravimetric analysis  
47 (TGA), ultraviolet-visible spectroscopy (UV-Vis) and microhardness measurements. The

1 results indicate that, although the recycling processes modify the structure of the  
2 nanocomposites, the recycled materials retain, to a great extent, the properties of the  
3 virgin material.

## 4 5 **2. Materials and methods**

### 6 **2.1. Materials and preparation of the samples**

7 The PLA used was a commercial grade designed for packaging (Ingeo 2003D,  
8 Natureworks™), with a melt mass-flow rate of 6 g/10 min (2.16 kg at 210 °C). Prior to  
9 processing, the material was dried according to the manufacturer's recommendations  
10 (20 minutes at 100 °C and 2 hours in a vacuum oven at 85 °C). The clay used was  
11 Cloisite™ 30B (C30), an organically modified montmorillonite supplied by Southern Clay  
12 Products (USA), where the organic modifier is a methyl, dihydroxyethyl, dehydrogenated  
13 tallow, quaternary ammonium chloride. The clay was dried in a convection oven at 100  
14 °C for 2 hours.

15 Neat PLA and the PLA nanocomposite with 2 wt.% C30 were melt compounded  
16 in a Rondol Microlab twin-screw microcompounder, with L/D = 20, at 60 rpm. The barrel  
17 temperature profile from hopper to die was 125,160,190,190,180 °C. The films with a  
18 thickness of  $230 \pm 10$  μm were obtained by compression in an IQAP-LAP hot-plate press  
19 at 190 °C, beginning with a melting step, with no pressure, for 5 minutes, followed by a  
20 degasification step for 2 minutes and cooling between cold plates at 14 MPa for 5  
21 minutes.

### 22 23 **2.2. Recycled materials**

24 In order to simulate the degradation of the material during its service life, the films  
25 of the virgin polymer were first subjected to an accelerated aging process, which included  
26 468 h of thermal degradation in a convection oven at 50 °C and 40 hours of  
27 photochemical degradation in an Atlas UVCON chamber, equipped with eight F40UVB  
28 lamps. A portion of these aged samples was washed at 85 °C for 15 minutes, in a solution  
29 of NaOH (1.0% by weight) and a surfactant (Triton X, 0.3% by weight), following the  
30 method proposed by Chariyachotilert et al. [24]. Finally, all samples were reprocessed  
31 by extrusion and compression in the above conditions.

32 According to the different recycling processes there were three different  
33 nanocomposites: PLAV-C30, which was not reprocessed, PLAR-C30, which was  
34 recycled without the washing step, and PLARW-C30 which was recycled with the  
35 washing step at 85 °C. Before the characterization, all the samples were subjected to  
36 physical aging at room temperature for 3 weeks.

### 37 38 **2.3. Experimental techniques**

39 Infrared spectra were recorded at a resolution of  $4 \text{ cm}^{-1}$ , with a total of 16 scans,  
40 using a Nicolet iS10 spectrometer equipped with a diamond Attenuated Total  
41 Reflectance (ATR) accessory.

1 The diffraction patterns of the materials were obtained using a X'PERT-MPD  
2 diffractometer, equipped with a CuK $\alpha$  generator ( $\lambda_1 = 0.154056$  nm and  $\lambda_2=0.154439$   
3 nm) at 45 kV and 40 mA, in a  $2\theta$  range from 1.7 to 50°. The basal spacing of the clay  
4 was calculated using the mathematical expression of Bragg's law, given in Eq. (1):

$$5 \quad \lambda = 2d \cdot \sin \theta \quad (1)$$

6 where  $\lambda$  is the wavelength of the rays,  $d$  is the spacing between the clay platelets and  $\theta$   
7 is the Bragg angle.

8 The intrinsic viscosity (IV) was measured in chloroform at  $25 \pm 0.5$  °C using an  
9 Ubbelohde viscosimeter. Intrinsic viscosity was obtained using the Kraemer equation  
10 [25], measuring four concentrations for each sample. The uncertainty was determined  
11 by linear regression with a 95% confidence level.

12 Differential scanning calorimetry (DSC) analysis was performed using a TA  
13 Instruments Q20 calorimeter on samples of 5-7 mg, in standard aluminum pans, at 5  
14 °C/min under nitrogen atmosphere. The samples were first heated from 20 to 200 °C,  
15 and then kept at 200 °C for 3 minutes to erase thermal history. After that, the samples  
16 were cooled to 0 °C, and finally a second heating scan was performed until 200 °C.

17 Thermogravimetric analysis was carried out using a TA Instruments TGA2050  
18 thermobalance. Samples of 12-14 mg were heated at 10 °C/min from room temperature  
19 to 800 °C in dry nitrogen (30 cm<sup>3</sup>/min).

20 The overall transmittance in the visible light region was measured according to  
21 the ISO 13468 standard, using a Shimadzu 2401 PC UV-Vis spectrophotometer  
22 equipped with a Shimadzu integrating sphere, using a scan speed of 200 nm/min.

23 The microhardness measurements were measured using a Type M Shimadzu  
24 microhardness tester equipped with a Vickers pyramidal indenter and applying a load of  
25 25 g for 10 seconds. Each measurement was repeated six times.

26 Water absorption was measured gravimetrically. Dried samples of 40 x 25 x 0.2  
27 mm were dipped in flasks containing 100 ml of 0.05 M phosphate-buffered solution (PBS)  
28 at pH  $7.4 \pm 0.2$ . The flasks were located in an oven set at 37 °C. Specimens were  
29 removed at selected immersion periods, gently wiped and weighed, at room temperature,  
30 in a laboratory balance with a precision of 0.1 mg. The percentage of absorbed water at  
31 any time  $t$ ,  $M_t$ , was determined by this expression:

$$32 \quad M_t(\%) = \frac{W_t - W_0}{W_0} \times 100 \quad (2)$$

33 where  $W_0$  and  $W_t$  denote, respectively, the weight of the specimens before and after the  
34 immersion in PBS.

35 The permeability measurements were performed in a homemade permeation cell  
36 that has been described elsewhere [26]. The device consists of a permeation cell with  
37 two chambers, separated by a PLA membrane of known thickness. Nitrogen, oxygen  
38 and carbon dioxide permeability of the different samples has been determined at 30 °C  
39 by means of diffusion experiments through an initially purged membrane. After purging, a  
40 0.2 MPa step variation of the pressure was imposed on the high-pressure side of the  
41 membrane, and the pressure on the low-pressure side was monitored. The permeability

1 coefficient, P, was estimated from the slope of pressure vs. time line after reaching  
2 steady state [27].

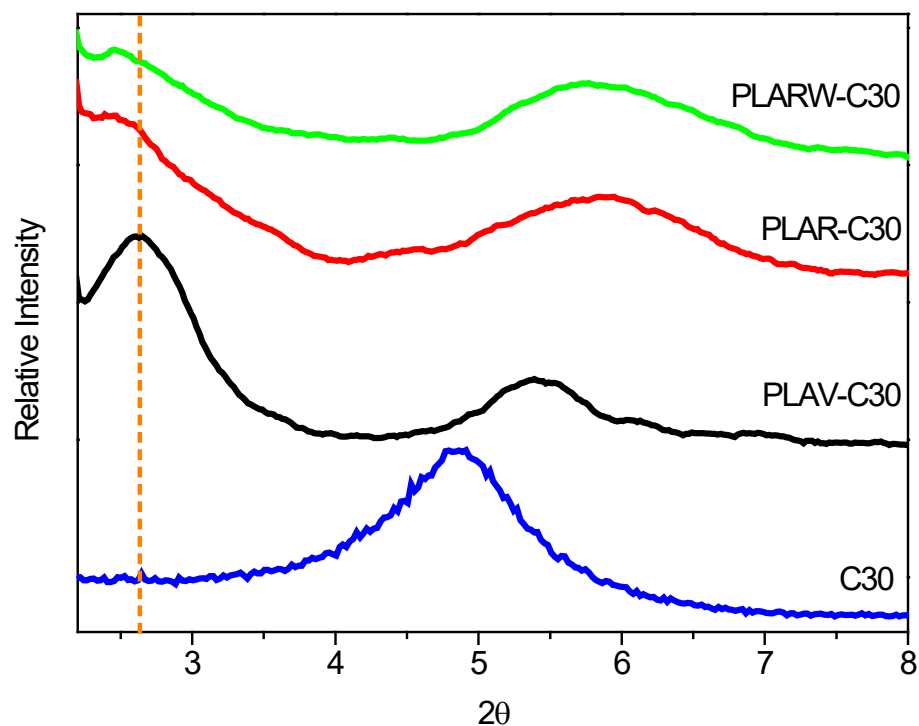
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### 4 3. Results and discussion

#### 5 3.1. Effects of recycling processes on the structure

6 The structure of PLA nanocomposites can be significantly altered by the  
7 degradation that takes place during the service life and the recycling process, which  
8 includes a melt reprocessing at high temperature and shear stress. On one hand, it is  
9 known that PLA and PLA-based materials are susceptible to thermal, photochemical and  
10 hydrolytic degradation processes that lead to a decrease of the average molecular mass,  
11 which can negatively affect the thermal, optical and mechanical properties of the material  
12 [28]. On the other hand, the dispersion of the nanoparticles in the polymer can be also  
13 altered during the recycling process due to the shear stress, resulting in a different  
14 morphology of the material, which also leads to changes in the properties.

15 The degree of dispersion was studied by means of XRD and TEM. Fig. 1 shows  
16 the diffraction patterns of the PLA-C30 nanocomposite subjected to the different  
17 mechanical recycling processes, along with the pattern of pure C30. There were no  
18 differences in the region between 8 and 20° (not shown), thus indicating that not  
19 important crystalline domains were formed in the polymer during the mechanical  
20 recycling of the nanocomposite. However, the changes observed at angles lower than  
21 8° reveal that the morphology of the composite was altered during the recycling  
22 processes.



23

24 **Fig. 1.** XRD patterns of pure C30 and PLA-C30 nanocomposite subjected to different  
25 mechanical recycling processes (V = virgin; R= recycled; RW = recycled with washing  
26 step).

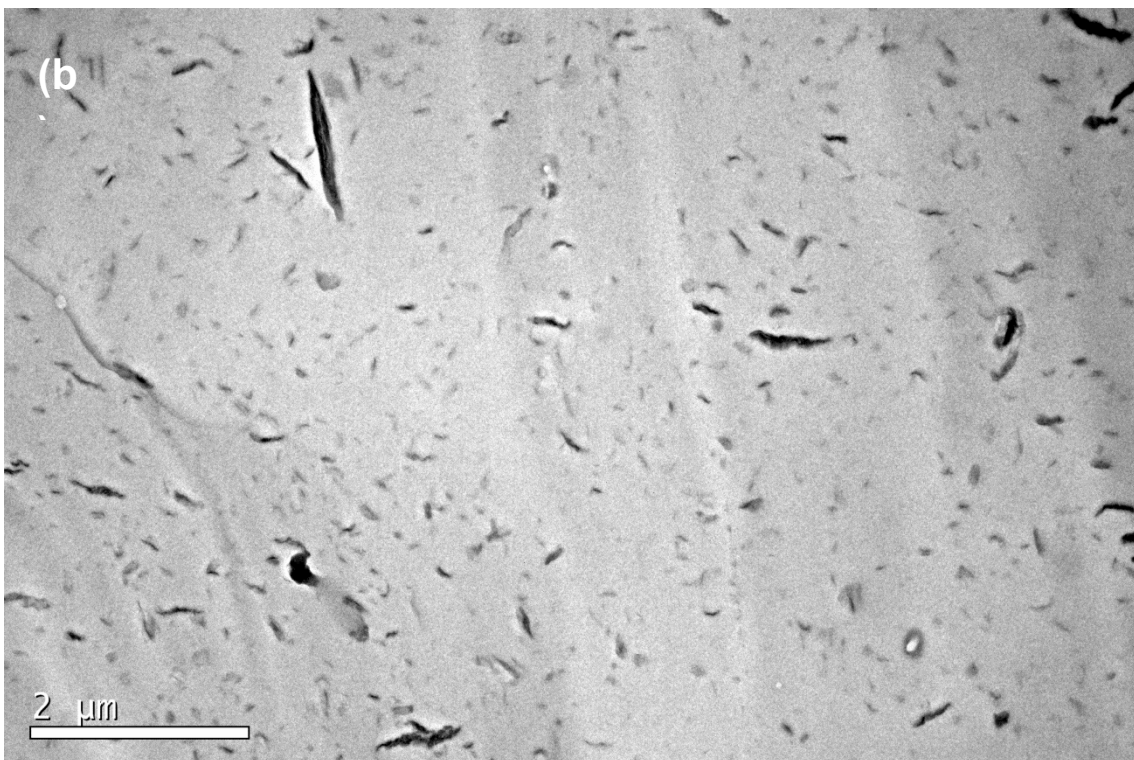
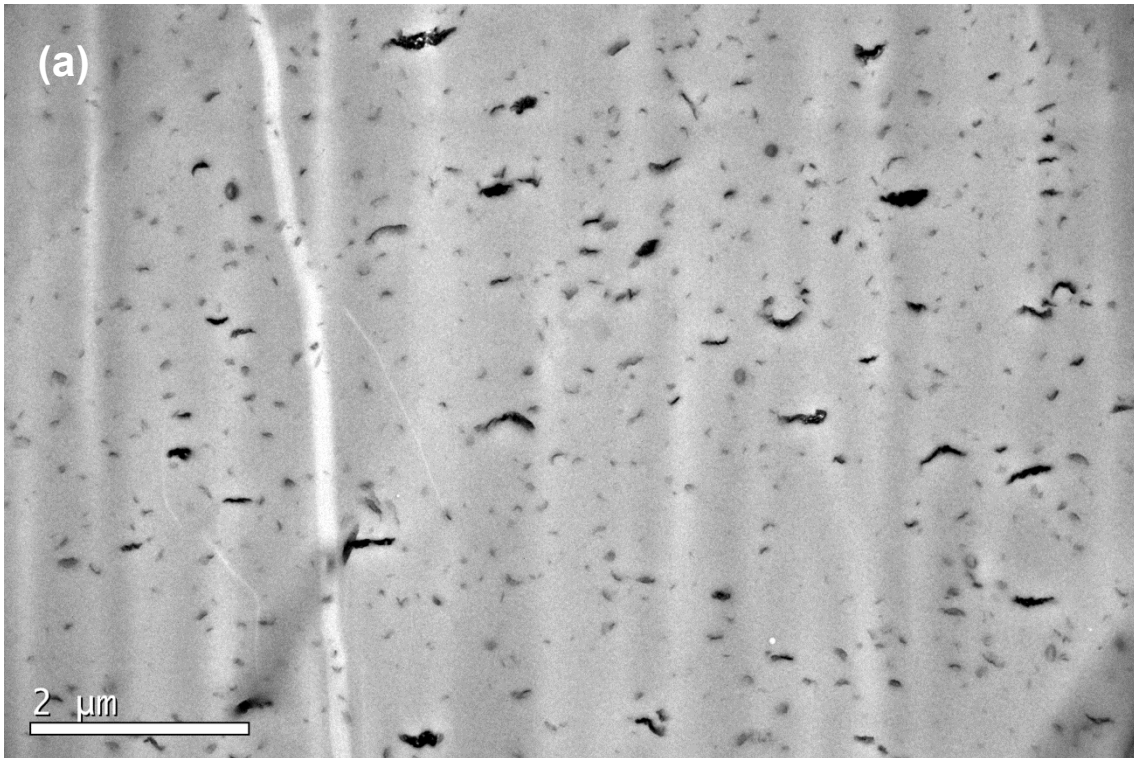
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1            Firstly, Fig. 1 shows the structure of the virgin nanocomposite. It can be seen that  
2 the characteristic (001) diffraction peak of the clay located at  $2\theta = 4.8^\circ$ , which  
3 corresponds to a basal spacing  $d_{001} = 1.8$  nm, is located at lower angles in PLAV-C30  
4 ( $2\theta = 2.6^\circ$ ), indicating the intercalation of the polymer chains within the galleries of the  
5 clay, and thus the formation of an intercalated structure. The nanocomposite also  
6 presents a broad second diffraction peak located at  $2\theta = 5.4^\circ$ , which may be related, at  
7 least partially, to the second-order reflection of the main diffraction. However, the  
8 formation of some clay aggregates with lower basal spacing during the melt processing  
9 could also contribute to this peak. A similar result was observed by Pluta et al. in PLA  
10 nanocomposites with 3 wt.% of C30 [29].

11            Regarding the effect of the different mechanical recycling on the degree of  
12 dispersion in the nanocomposite, Fig. 1 shows that the (001) peak disappears almost  
13 completely in PLAR-C30 and PLARW-C30. The remaining weak (001) diffraction is  
14 displaced towards lower angles, which can be attributed to an improved dispersion of  
15 the clay, forming exfoliated structures in which the individual silicate layers are no longer  
16 close enough to interact with each other. These results are very important, since an  
17 improvement of the clay dispersion could lead to enhanced mechanical, optical and gas  
18 barrier properties. These results agree with those obtained by Scaffaro et al. in PLA  
19 nanocomposites with 5% hydrotalcites, where an improvement of the clay dispersion  
20 was observed scanning electron microscopy (SEM) [23].

21            The effect of the recycling on the morphology can be also observed in the TEM  
22 micrographs corresponding to the virgin and recycled plastics (Fig. 2). It can be seen  
23 that, although some clay aggregates are still observed, the recycled material presents  
24 smaller particles than the virgin material, thus indicating an enhancement of the clay  
25 dispersion during the mechanical recycling, thus confirming the results observed in the  
26 XRD data.

27



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7

**Fig. 2.** TEM images of (a) PLAV-C30 and (b) PLAR-C30 (V = virgin; R= recycled).

The degradation of the polymer in the recycled materials was investigated by using intrinsic viscosity (IV) measurements, FTIR spectroscopy and differential scanning calorimetry.

1 The IV values are presented in Table 1. Firstly, it can be noticed that PLAR-C30,  
2 the nanocomposite recycled without the washing step, presents a reduction of around 7  
3 % on its intrinsic viscosity when compared with the virgin nanocomposite. This reduction  
4 in the intrinsic viscosity, and hence in the average molecular weight, is a consequence  
5 of chain scission processes in the polymer matrix due to the high temperatures and the  
6 shear stresses to which the polymer was subjected during the reprocessing, and could  
7 also be attributed to the degradation during the accelerated aging. Other authors have  
8 reported similar decreases in the molecular weight of unreinforced PLA samples  
9 subjected to a single reprocessing step [17,18,24]. This result is similar to that reported  
10 in neat PLA subjected to the same recycling process, in which a decrease of  
11 approximately 5% was obtained [1]. These results indicate that the clay does not have a  
12 very detrimental effect on the polymer degradation during the recycling of the  
13 nanocomposite. Moreover, the results reveal that the reduction of the intrinsic viscosity  
14 is small in this recycling process, so only limited effects on the mechanical and thermal  
15 properties of this recycled material must be expected.

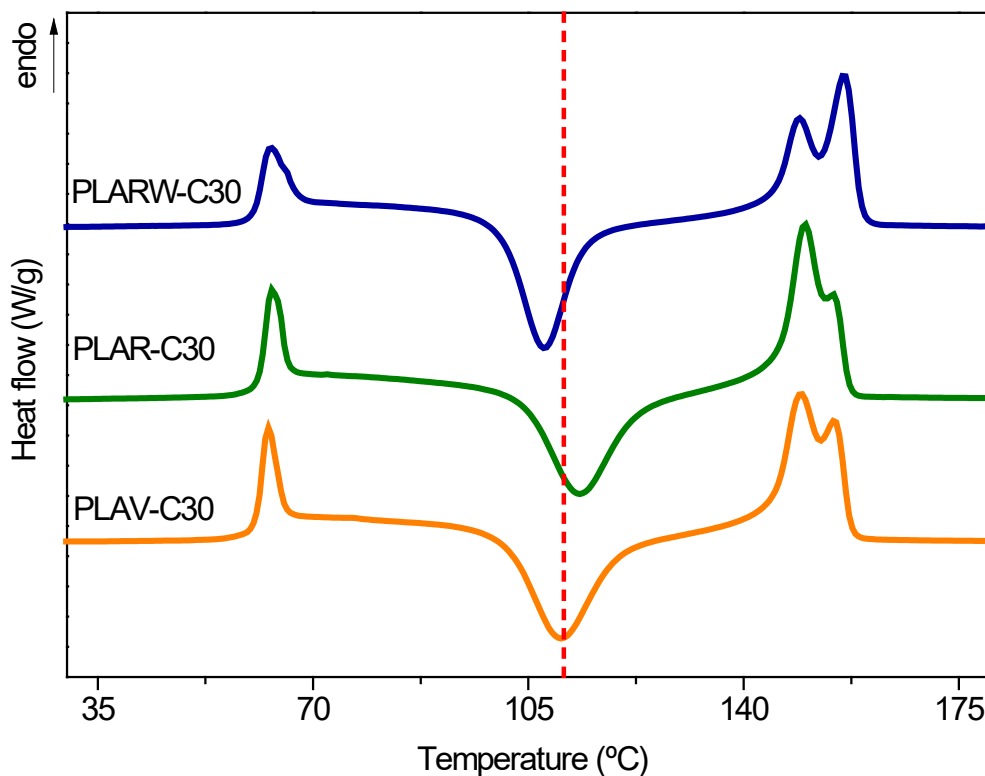
16 Regarding the material subjected to the recycling process that included a washing  
17 step, PLARW-C30, Table 1 shows that the decrease of IV is more important in this case  
18 (near 20 % when compared with the virgin nanocomposite). This result reveals that a  
19 demanding washing step such as the used in this work, with NaOH at high temperature,  
20 might play an important role on the degradation of PLA during its mechanical recycling,  
21 and thus could largely affect the properties of the recycled material. Again, unfilled PLA  
22 showed a very similar descent in the intrinsic viscosity, close to 20 %, when it was  
23 subjected to the recycling process including the washing step [1]. It was proposed that  
24 the washing in severe conditions weakened the structure of the polymer, generating acid  
25 groups that catalyzed the degradation during the following melt reprocessing.

26 The IV results indicate that the effects of the recycling strongly depend on the  
27 conditions selected for the recycling process. PLA-C30 nanocomposites can withstand  
28 a single recycling process without relevant decreases in the molecular weight of the  
29 polymer. However, the introduction of a demanding washing step in the recycling process  
30 promotes a further degradation of PLA during the reprocessing step, which might affect  
31 both the structure and final properties of the polymer.

32 After studying the effect of the different recycling processes on the molecular  
33 weight of PLA in the recycled nanocomposites with C30, the degradation suffered by the  
34 polymer was also studied by means of FTIR-ATR spectroscopy and DSC. The IR spectra  
35 (not shown) revealed that there are no major differences between the spectra of the  
36 studied materials. Although the IV values show the existence of chain scission during  
37 the recycling processes, which implies the formation of new carbonyl compounds, the IR  
38 spectra only show a minor difference in the carbonyl stretching band (centered at 1756  
39  $\text{cm}^{-1}$ ). This result is in good agreement with those reported by Badía et al. [18], which  
40 found that the generation of new carbonyl compounds during the reprocessing of PLA  
41 causes only a slight displacement of the carbonyl stretching band toward higher  
42 wavenumbers. Therefore, our results seem to indicate that, despite the degradation  
43 observed by means of intrinsic viscosity measurements, only small amounts of new  
44 carbonyl compounds are formed and the chemical structure of the polymer remains  
45 almost unchanged during the mechanical recycling.

1 DSC measurements can provide information about the structural changes of the  
2 polymer during the mechanical recycling of the nanocomposites. The first heating scans  
3 presented in Fig. 3 reveal that all the samples present a glass transition around 60 °C,  
4 followed by an endothermic peak related to the densification of the amorphous regions  
5 of PLA, that is, the physical aging of the polymer [20]. The samples also show a cold  
6 crystallization peak ( $T_{cc}$ ) at approximately 110 °C and finally, a double peak melting  
7 endotherm ( $T_{m1}$  and  $T_{m2}$ ) near 150 °C. The presence of this double melting peak has  
8 been explained as the result of a melt recrystallization process, which includes the  
9 melting of the less ordered crystals at a lower temperature ( $T_{m1}$ ), their reorganization into  
10 more perfect structures and their subsequent melting at a higher temperature ( $T_{m2}$ )  
11 [30,31].

12



13

14 **Fig. 3.** DSC first heating scans of nanocomposites subjected to different mechanical  
15 recycling processes (V = virgin; R= recycled; RW = recycled with washing step).

16

17 Concerning the recycling of the nanocomposite, it can be seen on Fig. 3 that  
18 PLAR-C30 shows a higher  $T_{cc}$  than the virgin nanocomposite, while PLARW-C30  
19 presents a lower value of  $T_{cc}$ . This behavior is different from that previously observed in  
20 the unreinforced polymer, where the mechanical recycling processes always led to lower  
21  $T_{cc}$  values in the recycled materials, which was explained as a consequence of the  
22 greater mobility of the shorter chains produced by the degradation during the recycling  
23 process [1,21]. The different behavior observed in the recycled nanocomposites  
24 suggests that the mechanical recycling presents in this case two opposing effects on the  
25 crystallization behavior of the polymer. On the one hand, the generation of shorter  
26 polymer chains facilitates the crystallization of PLA, while, on the other hand, the  
27 improvement of the dispersion of the clay platelets (observed by means of XRD) hinders

1 the packaging of the polymer chains, hence obstructing the formation of crystalline  
 2 domains. In the recycling without the washing step (PLAR-C30), the degradation was  
 3 small, as it was observed in the intrinsic viscosity measurements, thus prevailing the  
 4 hindering effect of the better dispersion of the clay. However, the degradation observed  
 5 when the demanding washing step was included (PLARW-C30) was significantly bigger,  
 6 counteracting the effect of the better dispersion of the clay and thus decreasing the  $T_{cc}$   
 7 of the material.

8 The presence of two opposite effects of the mechanical recycling in the  
 9 nanocomposites is also noticeable in the shapes of the melting endotherms of the  
 10 different materials. Fig. 3 shows that PLAR-C30 presents a larger low temperature  
 11 melting peak, which can be explained by the formation of less perfect crystalline structure  
 12 and the hindering of the melt recrystallization mechanism, caused by the better  
 13 dispersion of the clays in the polymeric matrix. On the other hand, PLARW-C30 shows  
 14 a larger high temperature melting peak, due to the presence of shorter polymer chains,  
 15 which can rearrange during the melting, forming more stable crystalline structures.  
 16 Despite the changes observed in the  $T_{cc}$  and melting endotherms, Fig. 3 shows that the  
 17 crystallization and melting enthalpies are similar for each material (the values, measured  
 18 as peak areas, are shown in Table 1), thus indicating that most of the crystals that melt  
 19 during the heating are previously formed in the cold crystallization. This behavior  
 20 suggests that all the materials are essentially amorphous, and that the different recycling  
 21 processes do not cause significant changes on the crystallinity of PLA, thus supporting  
 22 the results obtained from the XRD data. This conclusion is important, since the  
 23 development of crystalline structures could affect the mechanical, optical and gas barrier  
 24 properties of PLA. The fact that only marginal changes in crystallinity were observed  
 25 during the different mechanical recycling processes might imply that various important  
 26 properties of PLA remain unchanged despite the degradation of the polymer during the  
 27 mechanical recycling.

28  
 29 **Table 1.** Properties of the virgin and recycled nanocomposites (V = virgin; R= recycled;  
 30 RW = recycled with washing step). The given uncertainties were determined with a 95  
 31 % confidence level.

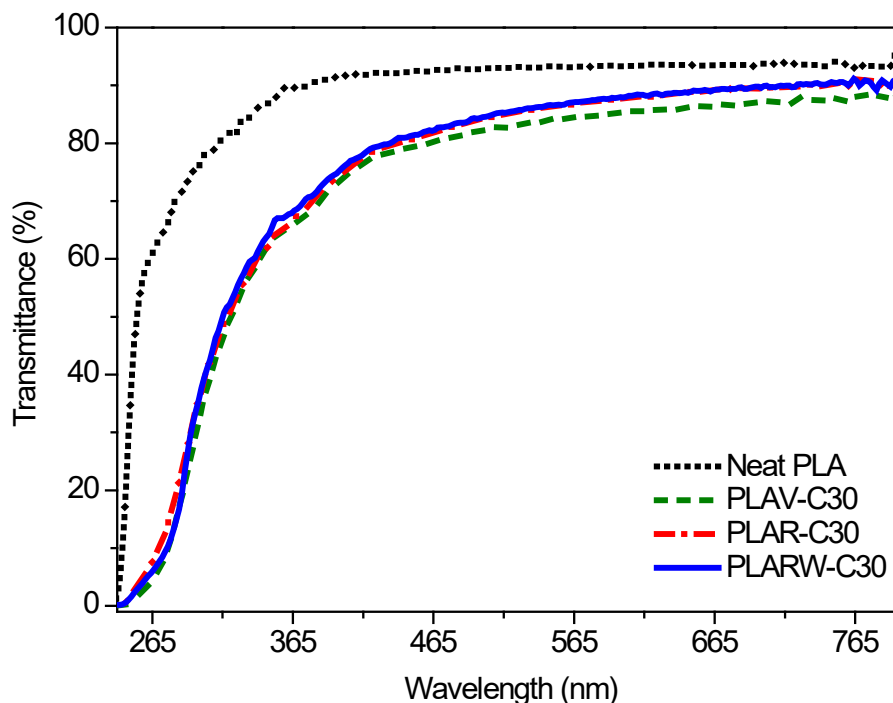
Material	Intrinsic viscosity (mL/g)	$\Delta H_c$ (J/g)	$\Delta H_M$ (J/g)	$T_{10}$ (°C)	$T_{max}$ (°C)	Visible light transmission (%)	Vickers hardness (MPa)
PLAV-C30	126 ± 3	27.0	29.2	340.0	367.3	83.4	189 ± 5
PLAR-C30	117 ± 4	25.9	28.3	338.7	377.7	85.6	184 ± 4
PLARW- C30	103 ± 4	27.7	28.2	339.4	374.1	85.9	180 ± 4

32  
 33 **3.2. Properties of the recycled materials**

1 After studying the effects of the different recycling processes on the structure and  
2 morphology of the nanocomposite with C30, the effects of the structural changes on  
3 some of the properties of the recycled materials were analyzed.

4 The thermal stability was studied by means of TGA. Table 1 gives the values of.  
5  $T_{10}$  and  $T_{max}$ , two characteristic temperatures commonly used as indicatives of the  
6 thermal stability of the materials, which are defined as the temperature at which 10% of  
7 the total mass is volatilized and the temperature of maximum rate of decomposition,  
8 respectively. Regarding the effect of the mechanical recycling on the thermal stability of  
9 the nanocomposite, it can be seen that both recycled materials present higher values of  
10  $T_{max}$ , despite the degradation observed in the intrinsic viscosity measurements. This  
11 behavior can be related at least in part to the better dispersion of the clay platelets,  
12 observed in the XRD measurements, which effectively act as a barrier for the liberation  
13 of the decomposition products of PLA, thus increasing the characteristic temperatures of  
14 the recycled nanocomposites. Furthermore, the chain scission of PLA during the  
15 recycling produces a decrease of the average molecular weight of the polymer, which  
16 reduces the thermal stability, but also the generation of new carboxyl groups, which leads  
17 to stronger interactions between the polymer chains and, hence, increased thermal  
18 stability of the recycled materials.

19 The changes in the morphology reported in the previous section can result in  
20 changes in the optical properties of the materials, which play a key role in the field of  
21 food packaging. The light transmission of the different materials was studied by means  
22 of UV-Vis spectroscopy, according to the ISO 13468 standard. The results are shown in  
23 Fig. 4 and Table 1.



24  
25 **Fig. 4.** UV-Vis spectra of the neat PLA and PLA-C30 nanocomposites subjected to  
26 different mechanical recycling processes (V = virgin; R= recycled; RW = recycled with  
27 washing step).

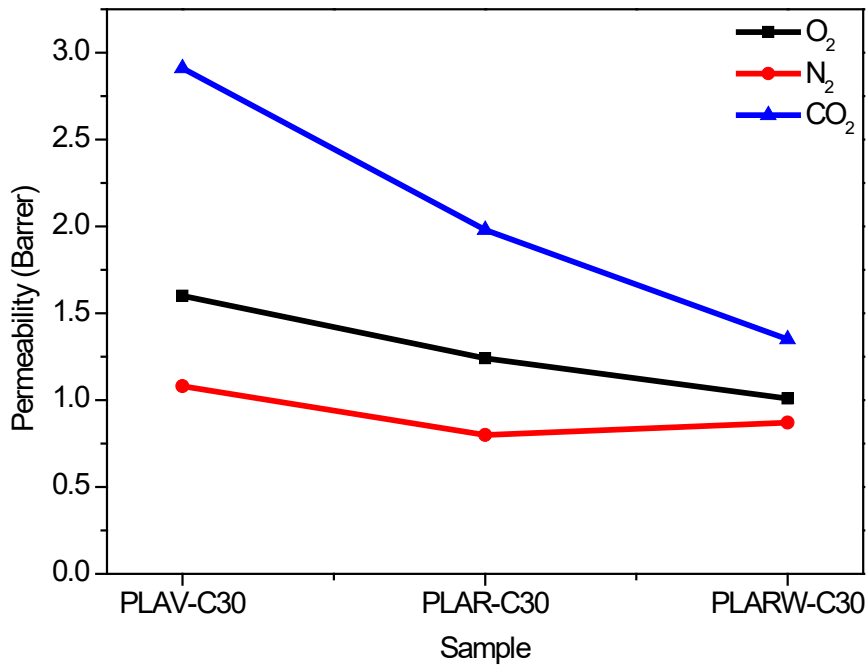
1 Fig. 4 shows the UV-Vis spectra of neat PLA, and the virgin and recycled  
2 nanocomposites. When compared with neat polymer, nanocomposites present higher  
3 absorption in the UV region of the spectrum, which confirms that the incorporation of C30  
4 provides some protection against the photochemical degradation caused by the UV  
5 radiation, which may be interesting from the point of view of food packaging applications.  
6 Respecting the visible part of the spectrum, between 400 and 800 nm, the addition of  
7 clays causes a decrease on the light transmission of the polymer, due to the presence  
8 of clay aggregates in the polymer matrix, which scatter and reflect the light, as it was  
9 pointed out by Cele et al. [32]. However, it is worth to note that both recycled  
10 nanocomposites present a higher visible light transmission than the virgin one. This  
11 result can be explained by the better dispersion of the clay nanoparticles in the recycled  
12 nanocomposites, observed by means of XRD and TEM, which reduces the amount of  
13 clay aggregates. Table 1 shows that the two recycled nanocomposites present good  
14 visible light transmission, higher than 85 %. In these materials, mechanical recycling not  
15 only does not worsen properties but even improves optical clarity.

16 The effect of the different recycling processes under consideration on the  
17 structure could also affect the mechanical properties and, hence, the potential use of  
18 recycled materials in various applications. In order to study the effects of the mechanical  
19 recycling, microhardness indentation tests were performed because it has been  
20 suggested that hardness measurement is more sensitive to molar mass changes than  
21 other mechanical characterization techniques [33]. The results that are shown on Table  
22 1 point out that the different mechanical recycling processes causes a small decrease,  
23 lower than 10%, in the Vickers hardness of the nanocomposite. The decrease of the  
24 hardness can be related to the lower molecular weight of PLA in the recycled materials  
25 [34]. However, the decrease is very small because the improved dispersion of the clay  
26 platelets in the recycled nanocomposites causes an increase on the hardness of the  
27 samples, thus partially counteracting the negative effect of the degradation of PLA during  
28 the different mechanical recycling processes. A similar behavior was observed by  
29 Scaffaro et al. in PLA/hydroxycalcites nanocomposites, which showed an increase of the  
30 Young modulus even after 3 reprocessing steps [23]. The moderate impact of the  
31 recycling processes on the hardness seems to indicate that the mechanical properties  
32 of the recycled materials should not be a limiting factor when considering the use of  
33 recycled PLA based materials in packaging applications.

34 The permeability is one of the most important properties of the materials used in  
35 food packaging applications, because this property is one the main factors for  
36 determining the shelf life of the packed products. In this work, the effect of the aging and  
37 recycling processes on the permeability of the material to O<sub>2</sub>, N<sub>2</sub> and CO<sub>2</sub> was evaluated.  
38 The results, shown in Fig. 5, indicate that the permeability of the nanocomposite clearly  
39 decreases as a result of the mechanical recycling. Again, this behavior must be  
40 explained by considering two opposing effects of the recycling processes. On one hand,  
41 the polymer degradation during the service life and the recycling process must favor the  
42 permeation, because fractional free volume,  $v_f$ , grows as a consequence of the molecular  
43 weight reduction and, according to Doolittle's equation, gas diffusion through a glassy  
44 polymer becomes easier if the  $v_f$  increases [35]. On the other hand, XRD and TEM  
45 experiments indicate that the melt reprocessing improves the dispersion of the  
46 impermeable clay platelets into the polymer, thus increasing the tortuosity of the paths  
47 followed by the gas molecules when diffusing through the polymer, which leads to

1 decreased permeability. In our case, the low permeability values corresponding to the  
2 recycled materials reveal that the improvement of the dispersion of the clay is the  
3 predominant effect of the recycling processes.

4



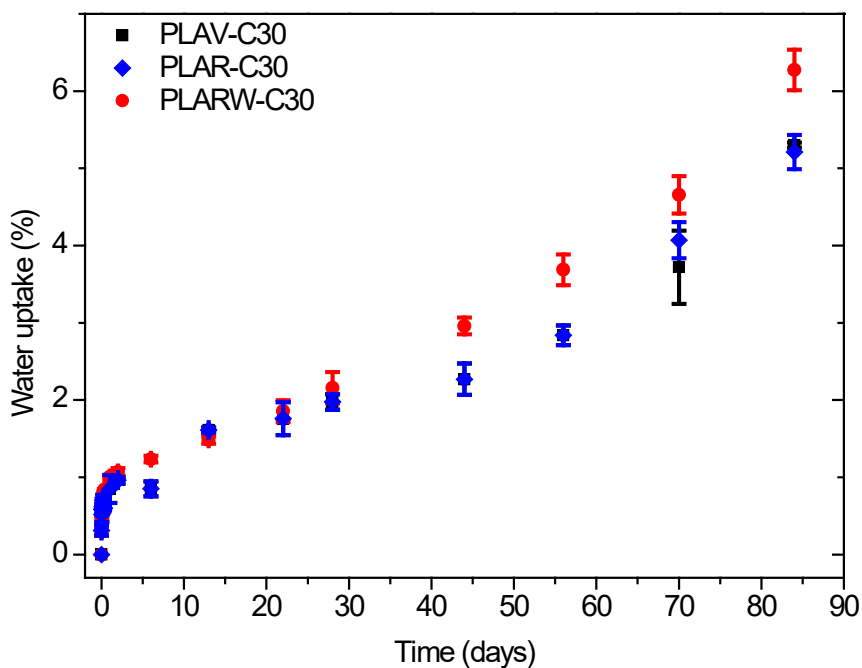
5

6 **Fig. 5.** Permeability of the virgin and recycled nanocomposites (V = virgin; R= recycled;  
7 RW = recycled with washing step).

8

9 In summary, the results indicate that the mechanical recycling of PLA-C30  
10 nanocomposites cause an improvement on the gas barrier properties as a consequence  
11 of the better dispersion of the clay due to the reprocessing.

12



13

1 **Fig. 6.** Water absorption curves for PLAV-C30, PLAR-C30 and PLARW-C30 ( $V =$   
2 virgin;  $R =$  recycled without washing step;  $RW =$  recycled with washing step).  
3

4 Water absorption and hydrothermal degradation tests are widely used to  
5 measure the quality and degradability of materials [36]. In addition, these tests are  
6 especially important in materials such as those studied in this work, because it is likely  
7 for these materials to be in contact with liquids or wet products during their service life.

8 Fig. 6 presents the water absorption curves of the virgin and recycled  
9 nanocomposites. It can be seen that for all materials, the amount of water absorbed does  
10 not reach equilibrium but grows continuously with the immersion time. This behavior was  
11 observed in a previous study with unfilled PLA [21], and was explained as the result of a  
12 two-stage sorption process. In the first stage, the fast water absorption is driven by the  
13 concentration gradient of the diffusant, while in the second stage the absorption is driven  
14 by the swelling and relaxation of the polymer [21]. Similar results have been reported by  
15 Davis et al. [21,37]. Furthermore, the hydrolytic degradation of PLA after long immersion  
16 time generates holes and hydrophilic products, which may facilitate the absorption of  
17 water [38]. Balart et al. also reported the appearance of cracks and holes in  
18 nanocomposites of PLA and hazelnut shell flour immersed in distilled water [36].

19 The complex nature of the sorption kinetics prevents the whole absorption  
20 process from being accurately described using a Fickian model, since this model  
21 assumes that the water transport is controlled by a concentration gradient and does not  
22 consider other phenomena such as the polymer degradation or the molecular relaxation.  
23 However, the first stage of the absorption process has great importance in practice for  
24 PLA-based materials, because an important fraction of these materials is used in the  
25 packaging of fresh food, so it would be interesting to elucidate if that first stage is  
26 controlled by a concentration gradient and can be accurately described using a Fickian  
27 model. An accurate model would allow the estimation of the water absorption  
28 parameters, and thus the study of the effect of the mechanical recycling on the kinetics  
29 of the first stage of the water absorption of the different samples.

30 For evaluating the accuracy of the Fickian model at short absorption times and  
31 calculating the parameters of the diffusion process, i.e., the diffusion coefficient and the  
32 moisture content at equilibrium, the experimental water absorption data for 120 minutes  
33 were fitted using a numerical solution of Fick's second law, given by Eq. (3) [39]:  
34

$$35 \frac{M_t}{M_\infty} = 1 - \frac{8}{\pi^2} \sum_{n=0}^{\infty} \frac{1}{(2n+1)^2} \exp\left(\frac{-(2n+1)^2 \pi^2 D t}{h^2}\right) \quad (3)$$

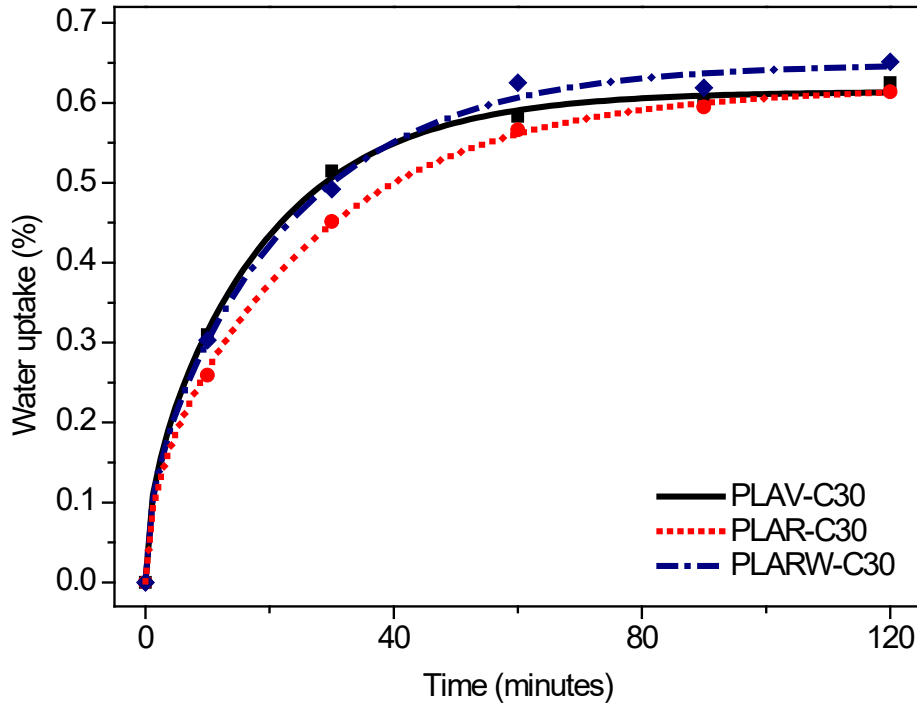
36  
37 where  $M_t$  is the moisture content at time  $t$ ,  $M_\infty$  is the moisture content at equilibrium or  
38 saturation mass,  $D$  is the apparent diffusion coefficient and  $h$  is the thickness of the  
39 sample. The serial expansion of Eq. (3) was truncated and only the first 10 terms were  
40 considered. Finally, the values of  $D$  were corrected by using Eq. (4), in order to consider  
41 the real dimensions of the samples used in the experiment, as indicated by Gupta and  
42 Pawar [40]:  
43

$$D_c = D \times \left(1 + \frac{h}{x} + \frac{h}{y}\right) \quad (4)$$

2

3 where  $D_c$  is the corrected diffusion coefficient and  $h$ ,  $x$  and  $y$  are the thickness, length  
 4 and width of the samples, respectively.

5



6

7 **Fig. 7.** Water absorption data, at short times, for the different samples. The lines  
 8 correspond to the fitting of the data using Eq. 3 (with  $n = 10$ ) (V = virgin; R= recycled;  
 9 RW = recycled with washing step).

10

11 The results obtained by fitting the experimental data using Eq. 3 are summarized  
 12 in Fig. 7 and Table 2. It can be seen that, for short immersion times, the water absorption  
 13 can be adequately fitted using Eq. 3, thus revealing that the diffusion of water is driven  
 14 by the concentration gradient in the first stages of the absorption.

15

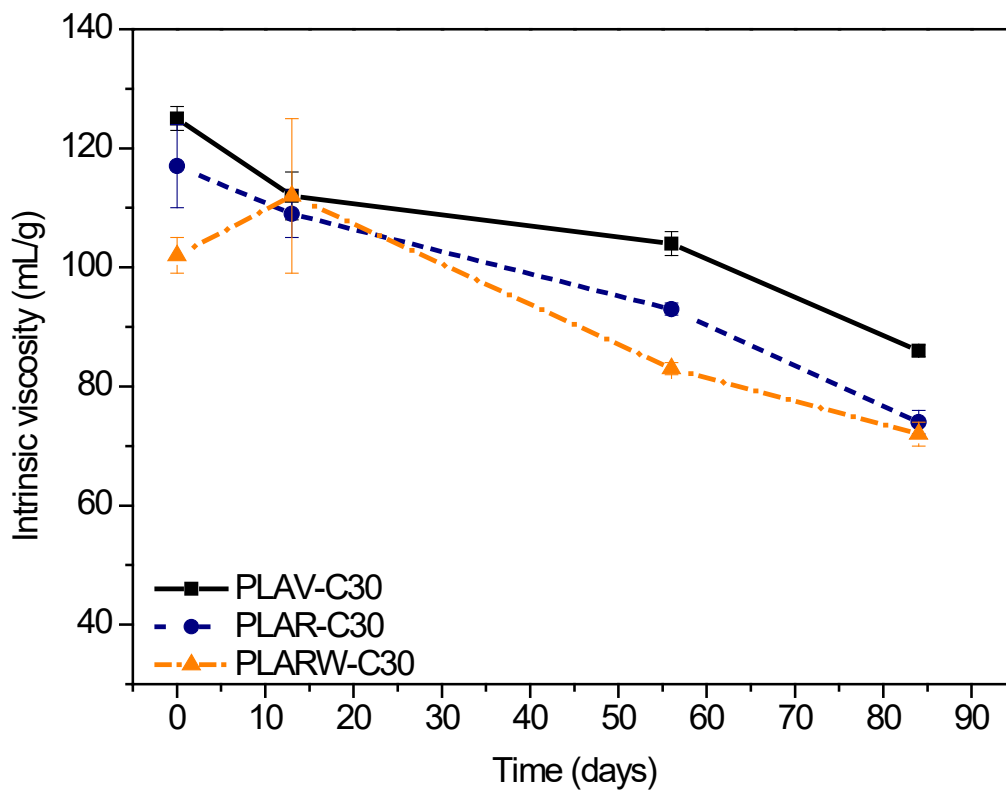
16 **Table 2.** Water absorption parameters calculated at short times for the different  
 17 materials using Eq. 3 (with  $n = 10$ )

Sample	$D_c \times 10^8$ (cm <sup>2</sup> /s)	$M_\infty$ (%)	$\chi^2$
PLAV-C30	$4.2 \pm 1.2$	$0.61 \pm 0.01$	0.0001
PLAR-C30	$3.8 \pm 1.5$	$0.62 \pm 0.01$	0.00003
PLARW-C30	$5.7 \pm 1.6$	$0.65 \pm 0.01$	0.0002

18

1 Table 2 and Fig. 7 reveal that virgin and recycled nanocomposites behave  
 2 similarly during the first stages of the absorption process. Only the material obtained  
 3 when the recycling process includes the washing step shows slightly higher values of  
 4 the diffusion coefficient and the saturation water. In order to explain these results, the  
 5 structural changes taking place during the recycling processes must be again  
 6 considered. The degradation of the recycled materials, responsible for the decrease in  
 7 the intrinsic viscosity that has been shown above, generates hydrophilic groups and an  
 8 increase of free volume that favors the diffusion and absorption of water. However, the  
 9 improved dispersion of clay nanoparticles slows the diffusion of water, so that recycling  
 10 causes a negligible net effect on the absorption of water at short times. The slightly higher  
 11 values of  $D$  and  $M_{\infty}$  in PLARW-C30 are explained as a consequence of the higher  
 12 degradation observed in this material.

13



14

15 **Fig. 8.** Evolution of the intrinsic viscosity of the materials during the immersion ( $V =$   
 16 virgin;  $R =$  recycled;  $RW =$  recycled with washing step).

17

18 The difference in water absorption between the nanocomposites is higher in  
 19 prolonged immersion times, as can be seen in Fig. 6. The increase in absorption at long  
 20 immersion times in PLA-based materials has been related not only to the degradation of  
 21 the polymer but also to a leaching process with the formation of pores and cracks [38].  
 22 Fig. 6 shows that there is a different behavior between PLAR-C30 and PLARW-C30. The  
 23 former shows a behavior very similar to that of the virgin nanocomposite, while the latter  
 24 shows a slightly higher absorption. These results indicate that the polymer degradation  
 25 during the accelerated aging, washing step and reprocessing favors the formation of  
 26 pores and cracks at long immersion times. To determine if the initial degradation of the

1 polymer also favors the decrease in the average molecular weight during the dipping,  
2 the intrinsic viscosity values were measured after different immersion times. The results  
3 shown in Fig. 8 indicate that the intrinsic viscosity decreases during immersion in water  
4 and that the decrease rate is similar in virgin and recycled nanocomposites.

5 The above results show that recycled nanocomposites have acceptable stability  
6 against hydrolytic degradation and confirm that recycled nanocomposites have good  
7 properties, comparable to those of the virgin material, so they could be used in  
8 demanding applications.

#### 10 **4. Conclusions**

11 The effects of accelerated aging and mechanical recycling processes on the  
12 structure and properties of PLA-clay nanocomposites have been studied. The results  
13 show two main differences between the structure of the recycled nanocomposites and  
14 that of the virgin. Firstly, mechanical recycling causes an improvement in the dispersion  
15 of the clay nanoparticles into the polymer, which has been observed by XRD and TEM.  
16 Secondly, intrinsic viscosity measurements reveal a decrease in the average molecular  
17 weight of the polymer in the recycled materials, especially when a demanding washing  
18 step is included during the recycling process. These two differences, which have also  
19 been observed in DSC curves, have opposite effects on the properties of recycled  
20 materials.

21 The virgin and recycled materials behave similarly in contact with water. The  
22 water absorption curves are complex, showing at least two distinct stages. In the first  
23 stage, at short immersion times, the absorption can be accurately described using a  
24 Fickian model. At long immersion times the absorption grows again due to the leaching  
25 and formation of voids and cracks in the polymer. The water absorption is slightly higher  
26 in the recycled material when the washing step is included in the recycling process, due  
27 to the higher molecular weight decrease in this case, which cannot be fully compensated  
28 by the better dispersion of the clay in the polymer. However, the decrease in intrinsic  
29 viscosity due to hydrolytic degradation is similar for virgin and recycled nanocomposites.

30 Despite the decrease of the molecular weight, the Vickers hardness remains  
31 almost unchanged and the thermal stability of the recycled nanocomposites is even  
32 slightly higher than that of the virgin, due to the better dispersion of the clay and the  
33 formation of carboxyl groups. The better dispersion of the clay also explains the higher  
34 optical clarity of the recycled nanocomposites, as well as the improved barrier properties  
35 for oxygen, nitrogen and carbon dioxide.

36 In general, the effect of accelerated aging and mechanical recycling processes  
37 on the properties of the nanocomposite is limited. The recycled nanocomposites have  
38 similar or even better properties in some cases than those of the virgin material, so the  
39 use of recycled nanocomposites in demanding applications, and even in the same  
40 applications as the virgin material, should not be ruled out.

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## 5 **6. Data availability**

6 The datasets analyzed during the current study are available from the  
7 corresponding author on reasonable request.

8

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